In seeking precedents for the Ag to Pt metal–metal bonding in this molecule, we have found only two very recently reported possibilities. B. L. Shaw et al.[5,6] have reported the preparation and structure of $(PhCC)_2Pt(\mu\text{-dppm})_2AgI$, with a Pt--Ag distance of 3.146 (3) Å, which the authors felt[5] did "not exclude significant bonding interaction". In the case of a trinuclear Pt–Ag–Pt compound the Pt–Ag distances of 2.849 (1) and 2.884 (1) Å were considered by Lippert and Neugebauer[7] to be "too long to implicate any strong metal–metal bonding...However, some weak metal–metal interaction appears feasible on the basis of Pt and Ag radii". Short, and hence potentially bonding, Ag–Ag distances have often been observed, and the range is from 2.740 (2) Å for a compound in which the silver atoms are bridged by a carbon atom of a ferrocene group[8] to distances as high as 2.957 (2)–3.085 (2) Å where an "attractive interaction between the Ag atoms" of an Ag_8 cube was proposed[9] and 3.074 (2) Å in a case where the authors neither proposed nor discounted Ag–Ag bonding.[10] The other reported Ag–Ag distances where bonding is considered to occur range from 2.757 to ca. 3.05 Å.[11-19] As further points of comparison, the Ag–Ag distance in metallic silver[20] is 2.889 (6) Å and pertinent sums of Pauling R_1 radii[4] are Ag–Ag = 2.68, Pt–Pt = 2.60, and Ag–Pt = 2.64. It may also be noted that the Ag–Rh bonds found in $\{[(\eta\text{-}C_5H_5)(CO)(PPh_3)Rh]_2Ag\}^+$ are 2.651 (1) and 2.630 (1) Å in length.[21]

Studies of the chemical reactivity of this unusual cluster anion are in progress and have revealed, inter alia, that with PPh_3 a scission occurs to give a $[PtAgCl_2(C_6F_5)_2PPh_3]^-$ ion in which, according to an as yet incomplete X-ray study, an Ag–Pt bond of length 2.80 Å is retained.

Acknowledgment. We thank the U. S. National Science Foundation and the C.A.Y.C.I.T. (Spain) for support.

Supplementary Material Available: A table of atomic positional parameters (2 pages). Ordering information is given on any current masthead page.

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Additions and Corrections

Orthoquinone Complexes of Vanadium and Their Reactions with Molecular Oxygen [*J. Am. Chem. Soc.* **1983**, *105*, 2680–2686]. Marion E. Cass, David L. Green, Robert M. Buchanan, and Cortlandt G. Pierpont*

Page 2680: The crystallographic unit cell dimensions in the abstract were presented in incorrect order. The cell constants should read as follows: a = 11.633 (4) Å, β = 13.102 (4) Å, c = 11.486 (4) Å, α = 108.81 (3)°, β = 97.82 (3)°, and γ = 100.58 (2)°.

Selectivities of π- and σ-Succinimidyl Radicals in Substitution and Addition Reactions. Appendix: Response to Walling, El-Taliawi, and Zhao [*J. Am. Chem. Soc.* **1983**, *105*, 5125–5131]. P. S. Skell,* R. L. Tlumak, and S. Seshadri

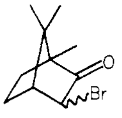
Page 5127, column 2, line 9: $k_{10}/k_7 = 6.9$ should be $k_{11}/k_8 = 6.9$.

Metacyclophanes and Related Compounds. 6. Reduction of [2.2]Metaparacyclophanequinone [*J. Am. Chem. Soc.* **1983**, *105*, 6650]. Masashi Tashiro,* Keizo Koya, and Takehiko Yamato

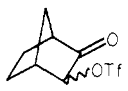
Page 6650: This statement "[2.2]metaparacyclophanequinones have not been synthesized previously" is an error. Earlier, Prof. H. A. Stabb et al. have prepared and characterized [2.2]metaparacyclophanequinone, which is published in *Tetrahedron Lett.* **1979**.

Solvolyses of α-Keto Norbornyl Trifluoroacetates and Triflates. Discrete α-Keto Cations vs. σ-Assisted (k_Δ) Processes [*J. Am. Chem. Soc.* **1983**, *105*, 7123–7129]. Xavier Creary* and Cristina C. Geiger

Reference to related studies on bromocamphor systems **a** was inadvertently omitted. While our rate and product studies implicated k_Δ processes in solvolyses of triflates **b**, previous studies on products formed from silver ion promoted reactions of **a** had also implicated k_Δ processes. See: (a) Bêquê, J. P.; Charpentier-Morize, M.; Pardo, C.; Sansoulet, J. *Tetrahedron* 1978, 293–298. (b) Charpentier-Morize, M. G. *Prepr., Div. Pet. Chem., Am. Chem. Soc.* **1983**, *28*(*2*), 297–318.



a



b

Photochemical Transformations. 35. Stereochemistry of Electron Transfer from Photoexcited Aromatic Rings to Carbon–Chlorine Bonds. Syn Sterochemistry of Migration in Photo-Wagner–Meerwein Rearrangements [*J. Am. Chem. Soc.* **1983**, *105*, 7337–7345]. Stanley J. Cristol,* Dave G. Seapy, and Ellen O. Aeling

Page 7343, column 1: In lines 22 and 29 2-(2′,3′-dimethoxybenzoyl)benzoic acid should be 2-(3′,4′-dimethoxybenzoyl)benzoic acid. In lines 28, 35, and 41 2-(2′,3′-dimethoxybenzyl)benzoic acid should be 2-(3′,4′-dimethoxybenzyl)benzoic acid.